Molar Mass Practice Answer Key

molar mass practice worksheet - northern highlands - c.p. chemistry name____ ws 4 more mole practice date____ pd.___ mass ↔ moles ↔ particles . molar mass practice *molar mass practice worksheet - my chemistry class* - solutions to the molar mass practice worksheet: important note to students: all of the units given here are "grams per mole", which may be abbreviated as "g/mol", "grams/mol", or "g . mol-1", depending on how your teacher likes to see it written. they all mean the same thing, but it's *mole review practice problems - rocklin.k12* - mole practice problems 1omic mass vs. molar mass: calculate the molar mass/ atomic mass of the following atoms and compounds: list atom types & #'s of each type of atom atomic mass (with correct units) molar mass (with correct units) h 2 o h= 2 o = 1 18.02 amu 18.02 grams/ mole fe 2 o 3 fe = 2 o = 3 159.70 amu 159.70 g/ mol nacl na = 1 cl = 1 *mole review practice problems - whitney high school* - mole practice problems i. atomic mass vs. molar mass: calculate the molar mass/ atomic mass of the following atoms and compounds: 1. complete the following table: list atom types & #'s of each type of atom atomic mass (with correct units) molar mass (with correct units) h 2 o h= 2 o = 1 18.02 amu 18.02 grams/ mole fe 2 o 3 nacl co 2 ba(oh) *chemistry computing formula mass worksheet - isd 622* - chemistry computing formula mass worksheet problem set-up example: find the formula mass of ca(no3)2 ... what is the molar ratio between o2and o3? practice problems ... chemistry stoichiometry practice(mass-molecule) answers: 1) 48 2) 2.39 x 1023 3) 1 x 1023 *molar mass worksheet - nclark* - molar mass worksheet calculate the molar mass of the following chemicals: 1) cl2 2) koh 3) becl2 4) fecl3 5) bf3 6) ccl2f2 7) mg(oh)2 8) uf6 9) so2 10) h3po4 11) (nh4)2so4 12) ch3cooh 13) pb(no3)2 14) ga2(so3)3 *molar mass worksheet answer key - riverdell* - "molar mass" is used to describe the mass of one mole of a chemical compound, while "atomic mass" is used to describe the mass of one mole of an element or the mass of one atom of an element. 8) which is a better unit for expressing molar mass, "amu" or "grams/mole"? *molar mass problems - chemteam* - molar mass problems 1. calculate the mass of 1.000 mole of cacl2 2. calculate grams in 3.0000 moles of co2 3. calculate number of moles in 32.0 g of ch4 4. determine mass in grams of 40.0 moles of na2co3 5. calculate moles in 168.0 g of hgs 6. calculate moles in 510.0 g of al2s3 7. how many moles are in 27.00 g of h2o 8.

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