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## Molar Mass Worksheet Answer Key

**molar mass worksheet - nclark** - molar mass worksheet calculate the molar mass of the following chemicals: 1)  $\text{Cl}_2$  2)  $\text{KOH}$  3)  $\text{BeCl}_2$  4)  $\text{FeCl}_3$  5)  $\text{BF}_3$  6)  $\text{C}_2\text{F}_2$  7)  $\text{Mg}(\text{OH})_2$  8)  $\text{UF}_6$  9)  $\text{SO}_2$  10)  $\text{H}_3\text{PO}_4$  11)  $(\text{NH}_4)_2\text{SO}_4$  12)  $\text{CH}_3\text{COOH}$  13)  $\text{Pb}(\text{NO}_3)_2$  14)  $\text{Ga}_2(\text{SO}_3)_3$  **molar mass worksheet - everett community college** - molar mass worksheet w 339 everett community college tutoring center student support services program find the molar masses of the following compounds: 1)  $\text{LiI}$  2)  $\text{PbCO}_3$  4)  $\text{Mg}(\text{OH})_2$  4)  $\text{K}_3\text{PO}_4$  5)  $(\text{NH}_4)_3\text{PO}_4$  6)  $\text{C}_6\text{H}_{12}\text{O}_6$  7)  $\text{Fe}_2(\text{SO}_4)_3$  8)  $\text{Na}_3\text{PO}_9$  9)  $\text{AgF}$  10)  $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$  **molar mass worksheet answer key - river dell regional ...** - molar mass worksheet ... "molar mass" is used to describe the mass of one mole of a chemical compound, while "atomic mass" is used to describe the mass of one mole of an element or the mass of one atom of an element. 8) which is a better unit for expressing molar mass, "amu" or "grams/mole"? ... **molar mass name chem worksheet 11-2** - what is its molar mass? 7. find the molar mass of acetylene  $\text{C}_2\text{H}_2$ , the gas used in welding torches. 8. vinegar contains acetic acid,  $\text{HC}_2\text{H}_3\text{O}_2$ . what is the molar mass of this substance? 9. what is the molar mass of magnesium sulfate,  $\text{MgSO}_4$ ? 10. find the molar mass of phosphorus pentachloride,  $\text{PCl}_5$ . 11. bromine is diatomic. find the molar ... **ws molar mass - saddleback college** - molar mass worksheet and key for the following compounds, write the chemical formula and determine the molar mass. write the units! water carbon dioxide sodium chloride calcium hydroxide potassium permanganate ( $\text{KMnO}_4$ ) sodium carbonate barium nitrate hydrogen monochloride sulfuric acid ( $\text{H}_2\text{SO}_4$ ) acetic acid ( $\text{C}_2\text{H}_4\text{O}_2$ ) see next page for key **molar mass practice worksheet - my chemistry class** - solutions to the molar mass practice worksheet: important note to students: all of the units given here are "grams per mole", which may be abbreviated as "g/mol", "grams/mol", or "g . mol<sup>-1</sup>", depending on how your teacher likes to see it written. they all mean the same thing, but it's **chemistry computing formula mass worksheet** - find the formula mass of the following compounds. round atomic masses to the tenth of a decimal place. place your final answer in the formula mass column. chemistry computing formula mass worksheet problem set-up example: find the formula mass of  $\text{Ca}(\text{NO}_3)_2$  ca:  $1 \times 40.1 = 40.1$  n:  $2 \times 14.0 = 28.0$  o:  $6 \times 16.0 = 96.0$  \_\_\_\_

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