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## Molar Ratio Worksheet Answers

**mole ratio worksheet - new york science teacher** - a) what is the  $\text{H}_2 / \text{H}_2\text{O}$  molar ratio? b) suppose you had 20 moles of  $\text{H}_2$  on hand and plenty of  $\text{O}_2$ , how many moles of  $\text{H}_2\text{O}$  could you make? c) what is the  $\text{O}_2 / \text{H}_2\text{O}$  molar ratio? d) suppose you had 20 moles of  $\text{O}_2$  and enough  $\text{H}_2$ , how many moles of  $\text{H}_2\text{O}$  could you make? 4) use this equation:  $\text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3$ , for the following problems **chemistry computing formula mass worksheet - isd 622** - the exact molar ratio you would use depends on how the problem is worded. what is the molar ratio between  $\text{O}_3$  and  $\text{O}_2$ ? what is the molar ratio between  $\text{O}_2$  and  $\text{O}_3$ ? practice problems following each equation are two requests for molar ratios from the equation. 1)  $\text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3$  write the molar ratios for:  $\text{N}_2$  to  $\text{H}_2$  and  $\text{NH}_3$  to  $\text{H}_2$  **stoichiometry worksheet 2 answer key - mrrromswinkel** - 4. given the following equation:  $\text{Na}_2\text{O} + \text{H}_2\text{O} \rightarrow 2 \text{NaOH}$  how many grams of  $\text{NaOH}$  is produced from  $1.20 \times 10^2$  grams of  $\text{Na}_2\text{O}$ ? how many grams of  $\text{Na}_2\text{O}$  are required to produce  $1.60 \times 10^2$  grams of  $\text{NaOH}$ ? 5. **stoichiometry practice worksheet - hazleton area school ...** - stoichiometry map for chemical reactions balanced chemical equation reactants products given grams wanted grams molar mass molar mass moles moles product  $x a y b + z c$  given: wanted: grams  $a \times 1$  mole  $a \times y$  mole  $b \times x$   $g b = \text{gram } b$  **molar mass worksheet answer key - river dell regional ...** - molar mass worksheet - answer key calculate the molar masses of the following chemicals: 1)  $\text{Cl}_2$  71 g/mol 2)  $\text{KOH}$  56.1 g/mol 3)  $\text{BeCl}_2$  80 g/mol 4)  $\text{FeCl}_3$  162.3 g/mol 5)  $\text{BF}_3$  **chemistry 11 mole fraction/molality worksheet date** - chemistry 11 mole fraction/molality worksheet name: date: 1. a solution is prepared by mixing 100.0 g of water,  $\text{H}_2\text{O}$ , and 100.0 g of ethanol,  $\text{C}_2\text{H}_5\text{OH}$ . determine the mole fractions of each substance. **mole calculation worksheet - sheffieldschools** - 1 mole = molar mass (could be atomic mass from periodic table or molecular mass) 1 mole = 22.4 l of a gas at stp (you do not need to worry about this yet) each definition can be written as a set of two conversion factors. they are: ... mole calculation worksheet ... **molar ratios and mass relationships in chemical equations - step 3:** calculate any molar masses (formula weights) needed and fill in the numbers using the balanced equation to find the molar ratios. molar mass = 0.0655 mole  $\text{CaCl}_2$  2 mole  $\text{NaCl}$  134.0 g  $\text{Na}_2\text{CO}_3$  4 4.39 g  $\text{Na}_2\text{CO}_3$  4 1 mole  $\text{Na}_2\text{CO}_3$  4 1 mole  $\text{Na}_2\text{CO}_3$  4 molar ratio the same basic steps work for all three problems. remember your ...

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